

An Overview of Bauxite Residue Utilisation

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Abstract



Bauxite residue is a solid waste generated during the production of alumina from bauxite. The disposal of this waste is a global challenge due to its potential environmental impact. If not well managed, its high alkalinity is a potential source of contamination of water, land and air in close proximity of the disposal site. Extensive work is carried out by researchers worldwide on value addition and fruitful utilization of bauxite residue. Some of the opportunities for utilization include adsorbents for the removal of heavy metals, dyes, phosphate, nitrate and fluoride; preparation of catalysts; recovery of iron, titanium and other trace metals; production of radio opaque materials; production of construction bricks; wood substitutes; cement; geo-polymers; development of coatings and pigments. This paper examines the details of bauxite residue utilization with the ultimate aim of solving the challenges of disposal and environmental impact.

Keywords: Bauxite residue utilization; wood substitutes; coatings and pigments; radio opaque material; geo-polymers

1. Introduction

Aluminium is a light weight, high strength structural material. Bauxite is the most economically important aluminium ore due to the presence of its high alumina content. Bauxite is a lateritic rock and contains mainly gibbsite, boehmite and diasporite. Bauxite also includes the iron oxides, goethite, hematite and small amounts of titanium, silica and other minerals and other impurities in minor or trace amounts. Bauxite usually contains the clay mineral kaolinite and has a density ranging from 2 600 to 3 500 kg/m³. Bauxite colour varies from whitish to pink to reddish brown depending on the iron content in the bauxite[1]. Globally, estimated bauxite resources are 55 to 75 billion tonnes, located in Africa (33 %), Oceania (24 %), South America and Caribbean (22 %), Asia (15 %) and elsewhere (6 %). India is self-sufficient in bauxite and has 3 billion tonnes of bauxite reserves out of global reserve of 65 billion tonnes. This places the country 5th in rank in the world [2]. Bauxite is classified based on ore type. These are mineralogically different and their occurrence in different countries is given in Table 1.

Table 1. Bauxite ore type in different countries [3].

| Gibbsitic | Boehmitic | Diasporic |
|--|---|--|
| Australia, Brazil, Guyana, India (Eastern Coast), Indonesia, Jamaica, Malaysia, Sierra Leone, Suriname, Venezuela. | Australia, Guinea, Hungary, USSR, Yugoslavia, India (Central part.) | China, Greece, Guinea, Romania, Turkey |

Aluminium production involves three stages: bauxite mining, followed by the refining of bauxite to alumina, usually in the Bayer process and smelting of alumina to aluminium by the Hall-Heroult process. Alumina can also be produced from bauxite under alkaline conditions using the lime sinter process [4], Deville Pechiney process (sodium carbonate) [5] and using the Serpeck process at high temperature in a reducing environment in presence of coke and nitrogen [6]. The Bayer process [7] is the most economic and prevalent method employed for extraction of alumina from bauxite, where it contains sufficient Al₂O₃ content. Bauxite of higher silica content is not suitable for the Bayer process and is more economically processed by the

sintering process. Waste generated from the sintering process is generally brown in colour, contains higher solid contents and lower quantities of Fe_2O_3 and Na_2O .

2. Bauxite Residue (Redmud)

In the Bayer process, the insoluble residue obtained after digestion of bauxite with sodium hydroxide at elevated temperature and pressure to extract alumina is known as red mud, or bauxite residue. Annual production of smelter and chemical grade alumina in 2015 was about 115 million tonnes produced using the Bayer process. The global average for the production of bauxite residue per tonne of alumina varies from 1 to 1.5 tonnes, so it is estimated that about 150 million tonnes of bauxite residue is produced annually. Bauxite residue has been called by different names by various companies. These are registered trademarks and reported in many publication and reports [8, 9]. Bauxite residue seems a more suitable name as some of the muds are brown rather than red and some diasporic derived residues are blackish. Despite this variation in colour, the name “red mud” remains more commonly used in North America and Europe [8, 9].

From an ecological viewpoint, whether it is red or brown mud is unimportant, both require appropriate waste and environmental management. The key risk with residue is the high alkali content, where pH values in fresh residue usually exceed 13. There is a risk of this alkalinity percolating into groundwater, in case of heavy rain and where appropriate monitoring and control measures are not in place. There is a risk of dam collapses (Hungarian alumina refinery, Ajka in 2010) and during the dry period of the year, dusting of the alkaline material may cause issues for local communities [10].

3. Environmental Consequences and Bauxite Residue Management

Bauxite residue produced from alumina refineries is a significant environmental risk for the surrounding environment including soil, vegetation and aquatic life, if not properly managed. The appropriate environmental management and/or utilization of bauxite waste are a major universal concern and preoccupation. The quantity and characteristics of the residue makes the task of safe disposal and utilisation more complex.

Safe storage of such large quantity of residue leads to the increase in production costs of aluminium by 2-5%. Residue storage also occupies large useful land tracts. The design and construction of such residue disposal facilities has varied considerably over the years with disposal practices generally dependent on nature of the immediate environment [11]. It is reported that conventional disposal methods involve the construction of clay lined dams or dykes, into which residue slurry is pumped and allowed to consolidate and dry naturally [12].

A good quantum of research and development work is being carried out all over the world in evolving a solution for the safe storage and utilization of bauxite residue. The current trend in residue storage is the increasing use of dry stacking as the preferred technology and further development in this area is being done to optimize the technology. Partial neutralisation using sea water is practiced at a number of Australian plants close to the sea; carbonation by using waste carbon dioxide from ammonia production has been adopted in Kwinana (Australia) and accelerated carbonation using intensive farming methods (Aughinish, Ireland), shows considerable benefits. Many of these approaches to disposal and management like alkali neutralisation using acid, CO_2 treatment, seawater neutralisation, bioleaching and sintering processes to fix leachables, stockpiling of bauxite residue alongwith their implications are discussed in detail elsewhere [3, 8].

Filtration using drum filters and plate and frame filter presses to recover caustic soda produce a lower moisture. This trend opens up benefits in terms of reuse with less than 28% moisture

The high pH is an issue from both a health and safety aspect and any adverse effects on a particular application product. The impact ranges from poor weather resistance in construction materials to high sodicity when used in soil amelioration. Both high sodium levels and high pH will be reduced when press filters are used. Carbonation using CO₂, intensive farming or acid neutralisation could also be considered. Based on a number of standard test criteria, material with a pH value above 12.5 is often considered as hazardous. Implementation of improved filtering operation, may reduce the pH to a level that avoids skin and eye irritation.

A high moisture level will increase transport costs, so it is beneficial for the bauxite residue to have as high solid content as possible. Additives such as starch have been used for dewatering, but from 1980, there has been a growing use of synthetic flocculants, although the use of plate and frame press filters (in combination with polymer based flocculation) is now being more widely adopted to reduce residue moisture content.

14. Conclusion

The comprehensive utilisation of bauxite residue generated in the process of industrial production of alumina is still a worldwide challenge. The current technologies and practices, the capacity for consumption and secondary utilisation are grossly insufficient to make an impression in the accumulation of bauxite residue inventories. Despite much work done over the last century, only some 2 - 3% of the bauxite residue produced annually is used in a productive manner. The largest future potential uses are as wood substitutes, geo-polymers, iron recovery, cement production, building materials, soil amelioration, landfill restoration and road construction materials.

It is to be noted that very large tonnages of equally hazardous industrial byproducts such as fly ash are routinely used in value-adding industrial applications. The same outcome should be possible for bauxite residue. The need is for bauxite residue producers, technology developers/providers, promoters, construction agencies, funding agencies, entrepreneurs, and users to work together for effective implementation of existing or new technologies and cater to the need, and fulfill the demand for sustainable development.

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16. References

1. *Geologydata.info*, Info portal of geology with special reference to Rajasthan, India. Available: [www.geologydata.info/bauxite deposits.htm](http://www.geologydata.info/bauxite%20deposits.htm).
2. *Geologydata.info*, Info portal of geology with special reference to Rajasthan, India. Available: [www.geologydata.info/bauxite deposits.htm](http://www.geologydata.info/bauxite%20deposits.htm).
3. Harekrushna Sutar et al. Progress of Red Mud Utilization: An Overview. *American Chemical Science Journal*, 4 (2014), 255-279.
4. D. C. Goldberg et al., Processes of extracting alumina from nonbauxite ores, Alkaline processes for low grade bauxites and clays, No. NMAb-278. *National Academy of Sciences-National Academy of Engineering*, Washington, D.C.; 1970.
5. Deville Pechiney process. Deville process. Available: [http://en.wikipedia.org/wiki/Deville process](http://en.wikipedia.org/wiki/Deville_process).
6. K.L.Reddy, Principles of Engineering Metallurgy. Extraction of non-ferrous metals. KK Gupta (Ed) ISBN: 81-224-0952,2001, *New Age International (P) Ltd.*, New Delhi, India, 3-35.

7. Bayer Process, Available: <http://en.wikipedia.org/wiki/BayerProcess>.
8. Kens Evans, Successes and challenges in the management and use of bauxite residue, *ICSOBA News Letter*, Vol. 14, 2015.
9. Kens Evans, *Aluminium International Today*, March/April 2016, 19-24. (www.aluminiumtoday.com).
10. www.intechopen.com
11. B. Solopek and J. Strazisar, The influence of red mud impoundments on the environment. *Light Metals* 1993, 41- 44.
12. R.Hind et al., The surface chemistry of Bayer process solids: a review, *Colloids and Surfaces A: Physicochem. Engg. Aspects*, 146 (1999), 359-374.
13. World Aluminium and the European Association, Bauxite Residue Management: Best Practice, <http://bauxite.world-aluminium.org/refining/bauxite-residue-management.html2015>.
14. W.D.Li, New Separation Technology Research of Iron from Bayer Progress Red Mud; *Central South University Library*: Changsha, China; 2006.
15. Y.F.Sun, FZ Dong andJT Liu, Technology for recovering iron from red mud by Bayer process (In Chinese), *Met. Mine*, 9 (2009), 176–178.
16. L.Zhong andYF Zhang, Sub molten salt method recycling red mud, *Chin. J. Nonferrous Met*, 18 (2008), 70–73.
17. X.F.Zheng, Recycling technology of aluminum and sodium from low temperature Bayer progress red mud, *Shandong Metall*, 32 (2010), 16–17.
18. XR Qiu andYY Qi, Reasonable utilization of red mud in the cement industry (In Chinese), *Chem. Techno.*, 6 (2011), 103–105.
19. E. Kalkan, Utilization of red mud as a stabilization material for the preparation of clay liners, *Eng. Geol*, 87 (2006), 220–229.
20. AA Barsherike, In *New Cement* (In Chinese); YY Qian. Ed. China Building Industry Press: Beijing, China; 1983.
21. I Vangelatos, GN Angelopoulos andD Boufounos. Utilization of ferroalumina as raw material in the production of Ordinary Portland Cement, *J. Hazard. Mater*, 168 (2009), 473–478.
22. Cement Research Institute of Construction Engineering. In *Red Mud Sulfate Cement* (In Chinese); *Metallurgy Industry Press*: Beijing, China. 1960.
23. ZH Pan, Research on Solid Activator-Slag and Red Mud Cement; *University of Nanjing Technology*: Nanjing, China; 1999.
24. NX Liang, DL Zhang andZX Yan, Study on performance of road made by cement red mud concrete (In Chinese), *China J. Highw. Transp*, 9 (1996), 6–11.
25. S Zhong, Red mud as admixture of concrete cements (In Chinese). *Jian Gong Ji Shu* 3 (1994), 17–22.
26. RB Hajela, RG Gupta andRK Goel, Disposal of solid wastes-red mud and fly-ash in the production of heavy clay products, *International Journal of Environmental Studies*, 33 (1989), 125-132
27. R Agarwal andM Shashikanth, Sintering Characteristics of Red Mud Compact. Thesis submitted in partial fulfillment of requirements for the degree of Bachelor of Technology in Metallurgical and Materials Engineering, National Institute of Technology, Rourkela, Odisha, India; 2008.
28. Y Pontikes, GN Angelopoulos and UL Kim, On the plasticity of clay mixtures with Bauxite residue of the Bayer process, *11th International Ceramic Congress and 4th Forum on New Materials*, Acireale, Sicily, Italy; 2006.
29. Y Pontikes et al., Effect of Firing Temperature and Atmosphere on Sintering of Ceramics Made From Bayer Process Bauxite Residue, *10th European Ceramic Conference and Exhibition*, University of Patras, Greece, Berlin; 2007.
30. G Xing andZZ Jiao, The development of non-autoclaved brick made of red mud and fly ash, *Rare Metals Cemented Carbide*, 6 (1993), 154–163.

31. AP Yang, The development of brick made of red mud and fly ash. *Light Metals*, 12 (1996), 17–18.
32. PX Zhang, Red mud making tile black granular materials (In Chinese). *Multipurp. Util. Miner. Resour*, 3 (2000), 41–43.
33. Y Nevin and S Vahdettin, Utilization of bauxite waste in ceramic glazes. *Ceram. Int*, 26 (2000), 485-493.
34. B Wu, DC Zhang and ZZ Zhang, The study of producing aerated-concrete blocks from red-mud (In Chinese), *China Resour. Compr. Util*, 6 (2005), 29–31.
35. S. C. Huang et al., Production of light weight aggregates from mining residue, heavy metal sludge and flyash, *Journal of Hazardous Materials*, 144 (2007), 52-58.
36. Mohini Saxena et al., Sisal fibre based polymer composites and their applications in a book on *Cellulose Fibers, Bio-, and Nano- Polymer Composites*, edited by Susheel Kalia, B. S. Kaith, Inderjeet Kaur, Publisher: (Springer-Verlag GmbH Berlin Heidelberg), ISBN 978-3-642-17369-1 , 4(22), (2011), 589-659.
37. Mohini Saxena et al., Composite Materials from Natural Resources: Recent Trends and Future Potentials, Pavla Těšinova(Eds.), *Advances in Composite Materials - Analysis of Natural and Man-Made Materials*, InTech, publisher, Croatia. ISBN 978-953-307-449-8. pp. 6:121-162, www.intechopen.com 2011.
38. Mohini Saxena, Prateek Mehrotra and P. Asokan, Innovative building materials developed from natural fibre and industrial waste, *Land Contamination & Reclamation*, 18(2011), 355-362.
39. Mohini Saxena, Potential of Bauxite Residue in Developing Innovative Materials for Engineering Application, *19th International Symposium ICSOBA*, 2012 in Belem, Brazil. P.43.
40. Mohini Saxena et al., Plant fibre-Industrial wastes reinforced polymer composites as a potential wood substitute materials, *Journal of Composite Materials*, 42 (2008), 367-384.
41. JZ Qi. Experimental Research on Road Materials of Red Mud, *University of Huazhong Science and Technology*: Wuhan, China; 2005.
42. JK Yang, F Chen and B Xiao. Engineering application of basic level materials of red mud high level pavement (In Chinese), *China Munic. Eng*, 5 (2006), 7–9.
43. LG Yang, ZL Yao and DS Bao. Pumped and cemented red mud slurry filling mining method (In Chinese). *Mining Res. Develop*, 16 (1996), 18–22.
44. HM Wang, The comprehensive utilization of red mud (In Chinese), *Shanxi Energy Conserv.* 11 (2011), 58–61.
45. G Zhang, J He and RP Gambrell, Synthesis, Characterization, and Mechanical Properties of Red Mud-Based Geopolymers, Transportation Research Record: *Journal of the Transportation Research Board*, 2167(2010), 1-9.
46. I Giannopoulou et al., Utilization of metallurgical solid byproducts for the development of inorganic polymeric construction materials, *Global NEST Journal*, 11 (2009), 127-136.
47. S.S. Amritphale et al., A novel process for making radiopaque materials using bauxite-Red mud, *Journal of the European Ceramic Society*, 27 (2007), 1945-1951.
48. W Huang et al., Phosphate removal from wastewater using red mud, *Journal of hazardous materials*, 158 (2008) ,35-42.
49. VK Gupta, I Ali and VK Saini, Removal of chlorophenols from wastewater using red mud: an aluminum industry waste, *Environmental Science & Technology*, 38 (2004), 4012-4018.
50. A Tor et al., Removal of phenol from aqueous phase by using neutralized red mud. *Journal of Colloid and Interface Science*, 300 (2006), 498-503.
51. A Tor, N Danaoglu and Y Cengeloglu. Removal of fluoride from water by using granular red mud: Batch and column studies. *Journal of hazardous materials*. 164 (2009). 271-278.

52. Y Cengeloglu et al. Removal of boron from aqueous solution by using neutralized red mud, *Journal of hazardous materials*, 142(2007), 412-417.
53. VK Gupta, M Gupta and S Sharma, Process development for the removal of lead and chromium from aqueous solutions using red mud-an aluminium industry waste, *Water Research*, 35 (2001), 1125-1134.
54. VK Gupta and S Sharma, Removal of cadmium and zinc from aqueous solutions using red mud, *Environmental Science & Technology*, 36 (2002), 3612-3617.
55. C Brunori et al., Reuse of a treated red mud bauxite waste: studies on environmental compatibility, *Journal of Hazardous Materials*, 117 (2005), 55-63.
56. SJ Palmer and LF Ray, Characterisation of bauxite and seawater neutralized bauxite residue using XRD and vibrational spectroscopic techniques, *Journal of Material Science*, 44 (2009), 55-63.
57. Sara J Palmer et al., Thermally activated seawater neutralised red mud used for the removal of arsenate, vanadate and molybdate from aqueous solutions, *Journal of Colloid and Interface Science*, 342 (2010), 147-154.
58. Z Shuwu et al., Arsenate removal from aqueous solutions using modified red mud, *Journal of Hazardous Materials*, 152 (2008), 486-492.
59. GH Fuhrman, CT Jens and D McConchie, Adsorption of arsenic from water using activated neutralized red mud, *Environmental Science & Technology*, 38 (2004), 2428-2434.
60. E Fois, A Lallai and G Mura, Sulfur dioxide absorption in a bubbling reactor with suspensions of Bayer red mud, *Ind. Eng. Chem. Res*, 46 (2007), 6770-6776.
61. Y Chen, JQ Li and F Huang, The Performance Research on Absorbing SO₂ Waste Gas with Bayer Red Mud (In Chinese), *J. Guizhou Univ. Technol. Nat. Sci.*, Ed. 36 (2007)30-37.
62. AI Cakici et al., Utilization of red mud as catalyst in conversion of waste oil and waste plastics to fuel, *Journal of Materials Cycles and Waste Management*, 6 (2004), 20-26.
63. D Garg and EN Givens, Coal Liquifaction catalysis by industrial metallic wastes, *Ind. Eng. Chem. Process Des. Dev.*, 24 (1985), 66-72.
64. JS Shing. Method of activation of red mud. *US Patent* 4017425. 1977.
65. W Shaobin, HM Aug and MO Tade, Novel applications of red mud as coagulant, adsorbent and catalyst for environmentally benign processes, *Chemosphere*, 72 (2008), 1621-1635.
66. A Satapathy et al., Red mud: A potential coating material, *National seminar on environmental concern and remedies in Alumina Industry at NALCO*, Damanjodi, India, 2007, 139-142.
67. A Satapathy et al., Characterization of Plasma Sprayed Pure Red Mud Coatings: An Analysis, *American Chemical Science Journal*, 3 (2013), 151-163.
68. H Sutar et al., Morphology and solid particle erosion wear behavior of red mud composite coatings, *Natural Science*, 4(2012), 832-838.